

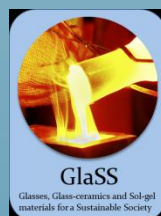
Crystallization and optical properties of Rare earth doped-oxyfluoride glass-ceramics for its use in optical fibers

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Outline

Introduction

Glass and Glass-ceramics (GC) preparation:



Thermal and Structural Characterization

Optical Properties

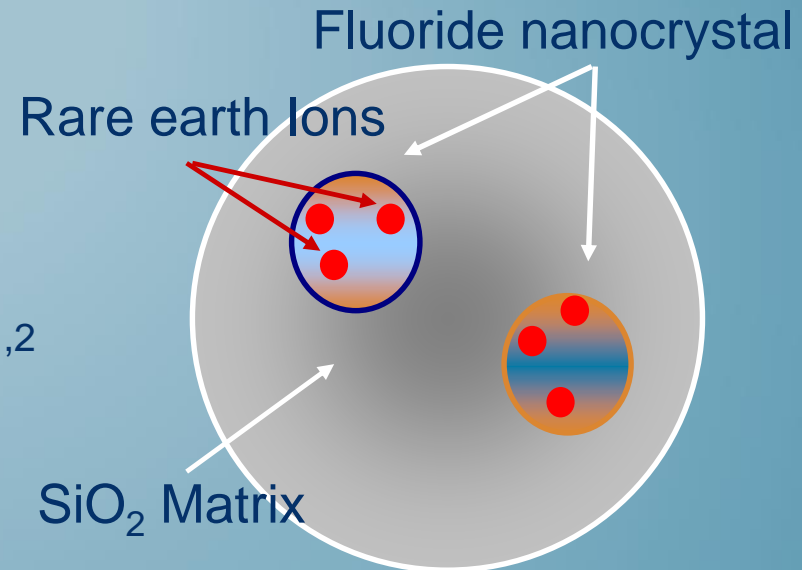
Conclusions

Acknowledgements

Why oxyfluorides glass-ceramics doped with Rare earths ions?

- Enhanced the mechanical properties
- Enhanced the optical properties

(low phonon environment ($300\text{-}400\text{ cm}^{-1}$))^{1,2}



These nanoglass-ceramics combine the optical properties of Rare earths ions in fluoride system with the advantages of elaboration and use of oxides^{3,4}

¹Wang, Ohwaki., App.Phys.Lett, 63, 3268 (1993)

²Fedorov, P.P. Luginina, A., Popov, A.I. J Fluorine Chem. 172, 22-50, (2015)

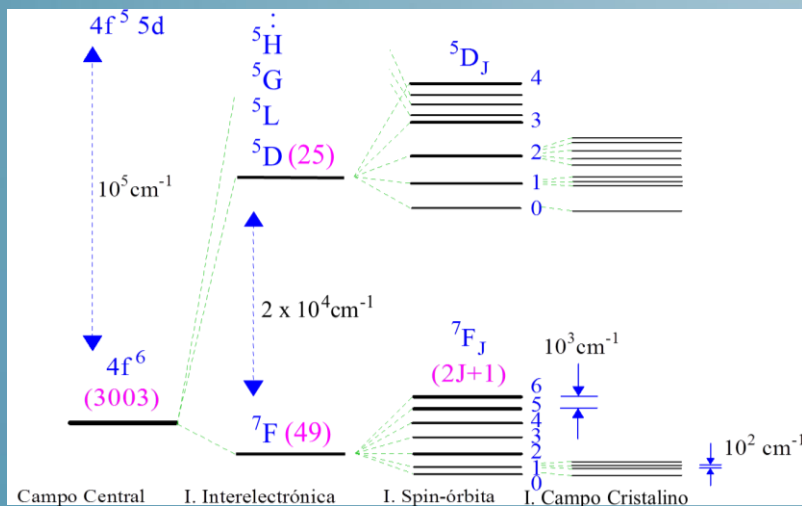
³de Pablos Martín A. et al : J. Am. Ceram. Soc. 94, [8], 2420-2428 (2011)

⁴de Pablos-Martín A., Durán A., Pascual M. J.: Int. Mater. Rev. 57 [3], 165-186 (2012)

Introduction

Why oxyfluorides glass-ceramics doped with Rare earths ions?

- ✓ 4f electrons of Rare earths ions are partially shielded from the external interactions due to the existence of external electrons (5s, 5d)

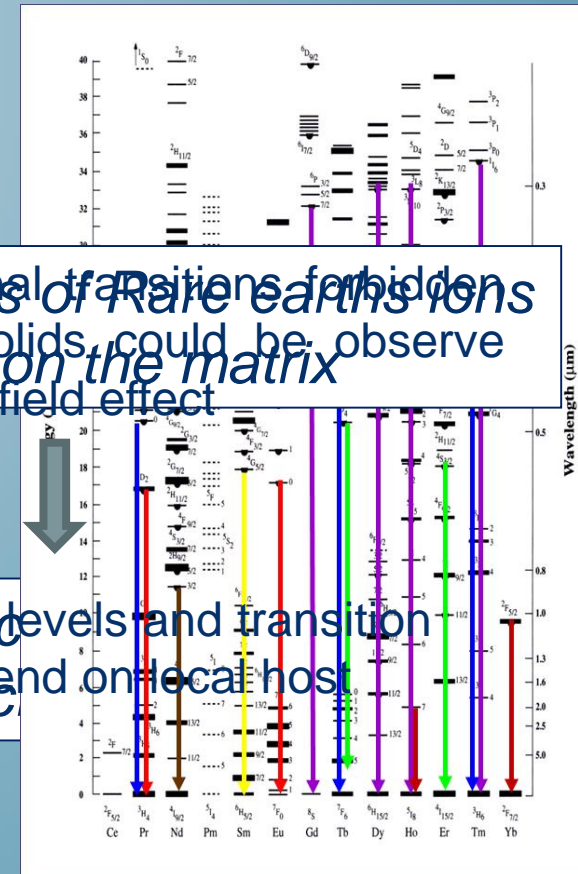


✓ Intradonfigurational transitions forbidden in free ion, in solids could be observe due to crystalline field effect

Optical properties of Rare earths ions depends on the matrix

✓ Splitting of Stark levels and transition probabilities depend on local host

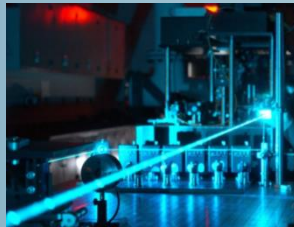
More efficient nanoc



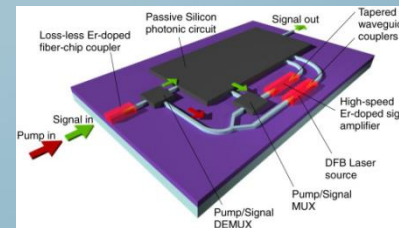
Introduction

Applications

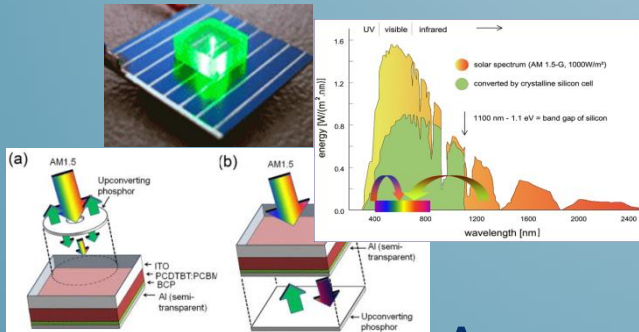
- Fluorescent markers



- Active media on lasers emitters

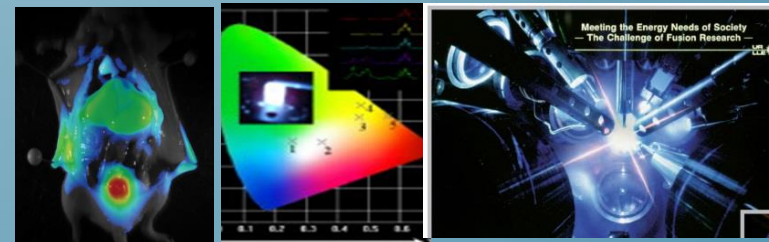


- Integrate photonic devices



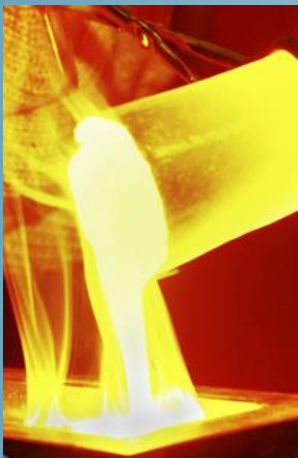
- Enhanced the efficiency of solar cells through photon-conversion processes

- Among others



Glass and Glass-ceramics preparation

$70\text{SiO}_2\text{-}7\text{Al}_2\text{O}_3\text{-}8\text{Na}_2\text{O-}8\text{K}_2\text{O-}7\text{GdF}_3\text{-}x\text{ErF}_3\text{-}y\text{YbF}_3$ (% mol)
(70Si-7Gd)



1600°C during 1h \longleftrightarrow 1650°C during 0.5h
515 °C during 0.5h

Re^{3+} and $\text{Re}^{3+}\text{-Yb}^{3+}$ doped 70Si-7Gd glasses with the following Re^{3+} concentrations (in mol%) were prepared:

- undoped
- 0.5 Er^{3+}
- 0.5 $\text{Er}^{3+} - 2\text{Yb}^{3+}$

1 Glass

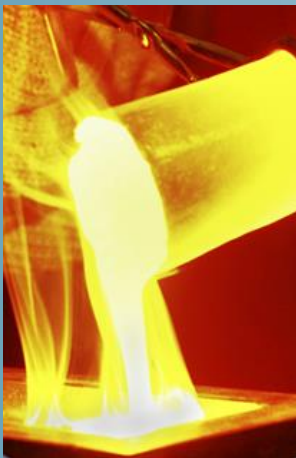
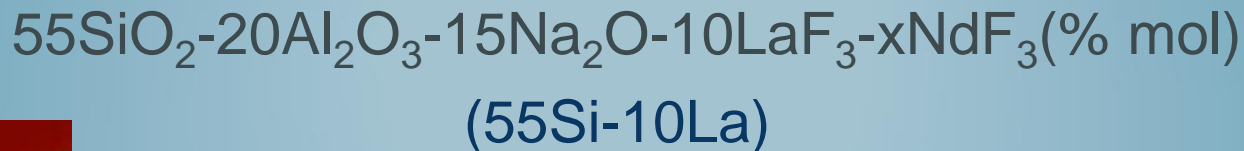
2 Glass-ceramics:

550°C-80h

580 °C-120h



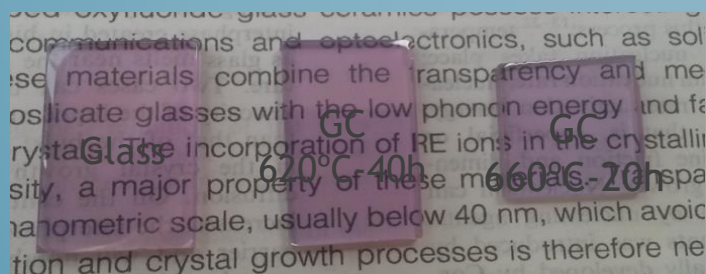
Glass and Glass-ceramics preparation



1650°C during 1.5h \longleftrightarrow 550 °C during 0.5h

Re³⁺ doped 55Si-10La glasses with the following Re³⁺ concentrations (in mol%) were prepared:

- undoped
 - 0.1, 0.2, 0.5, 1, 2 Nd³⁺
- 1 Glass
- 2 Glass-ceramics: 620°C 1÷80h
660°C 20h



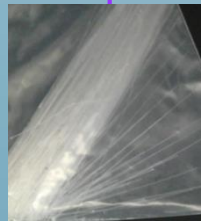
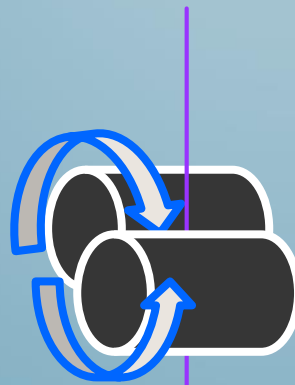
Glass and Glass-ceramics preparation: Fibers

Fibers 55Si-10La

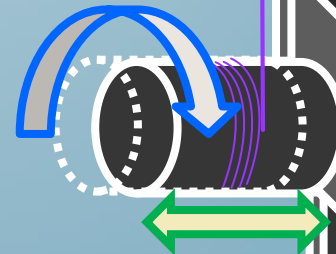


1480°C

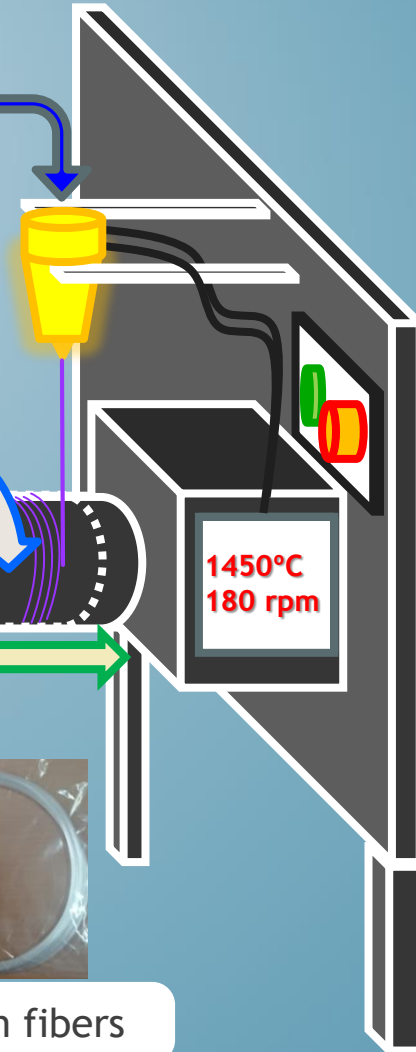
$\log \eta \sim 2.8$; GCs 640°C 40h



150-500 μm fibers



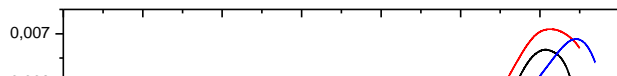
20-80 μm fibers



Thermal Characterisation: Dilatometry and DTA

70Si 7Gd

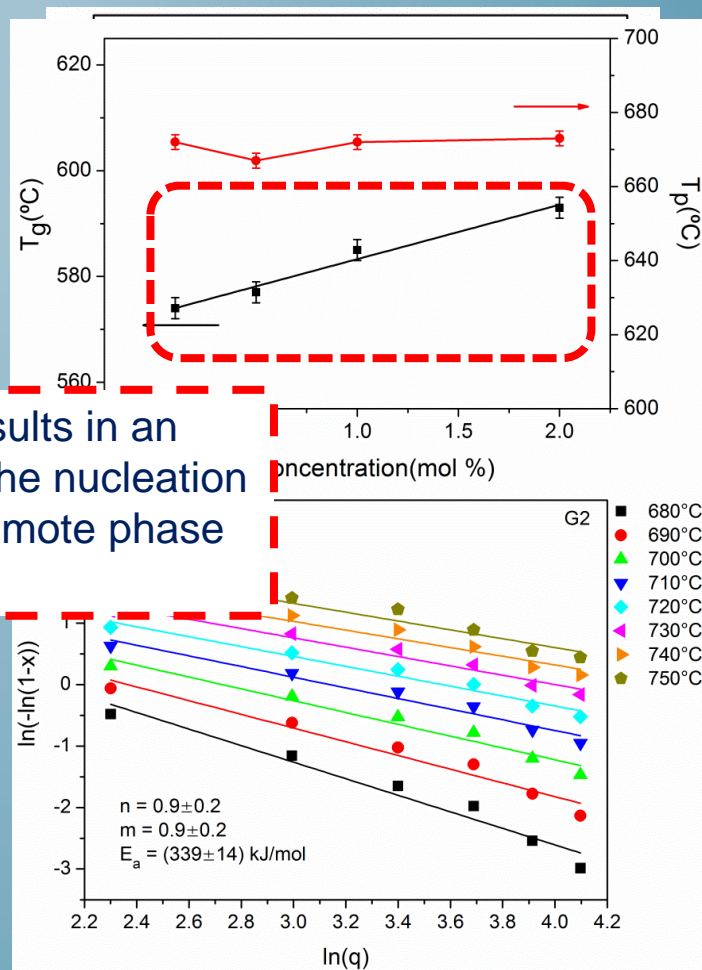
55Si 10La



	Undoped	0.5 Er ³⁺	0.5 Er ³⁺ - 2Yb ³⁺
T_g (°C) ± 3	510	530 ∇	546
T_s (°C) ± 6	590		
$\alpha \cdot 10^{-6}$ (K ⁻¹) ± 0.5	10.2		

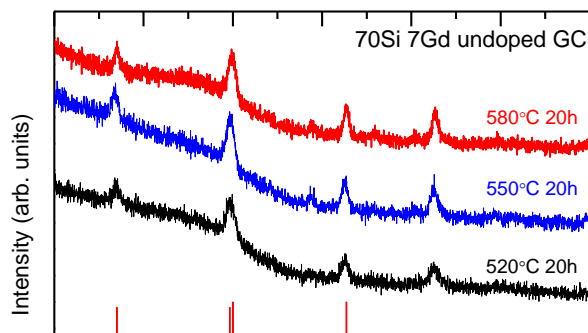
Dopant incorporation results in an increment of the T_g , due to the nucleation action of Re^{3+} ions that promote phase separation

Crystallisation mechanism controlled by diffusion that starts from a constant number of nuclei and the crystals growth is in 2D - nanoplates



Structural characterization: XRD

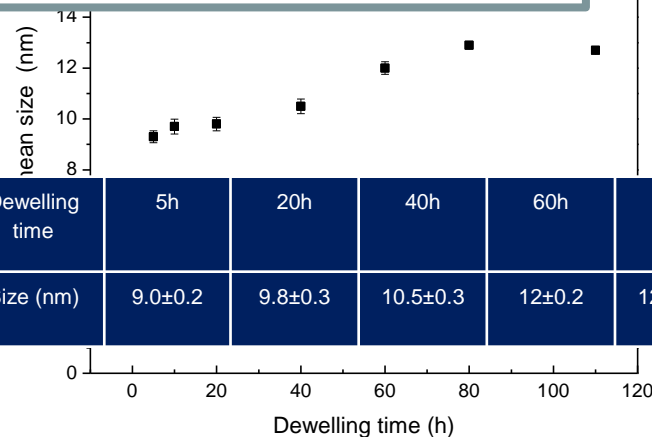
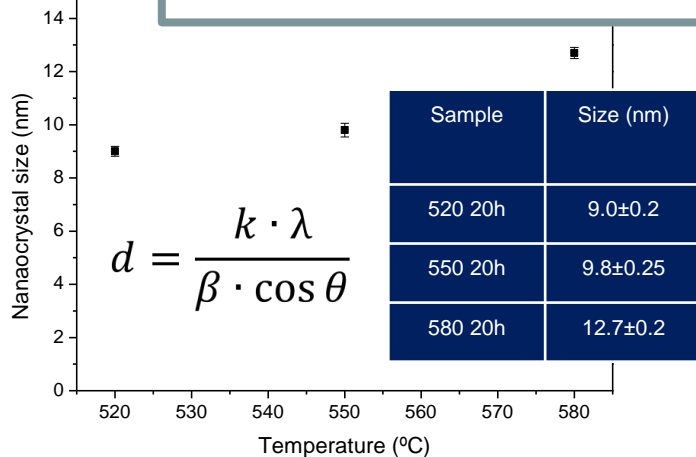
70Si-7Gd



Intensity (arb. units)

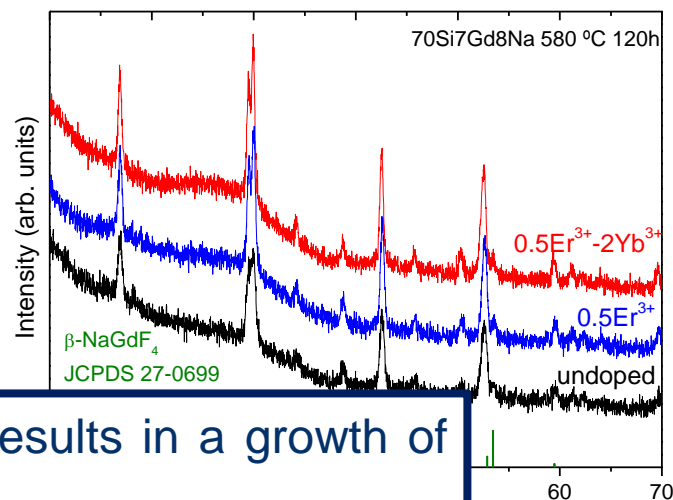
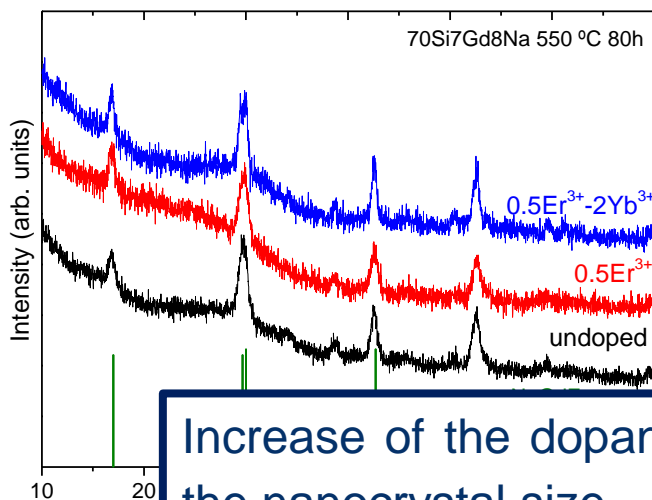
After heat treatment at 520, 550 and 580 °C phase β -NaGdF₄ (JCPDS 27-0699) have been obtained.

Increase of dwelling time at 550 °C, nanocrystal mean size growth up to 13.0 nm. This nanocrystal size remains constant after 80 h at this temperature.



Structural characterization: XRD EAGLES

70Si-7Gd

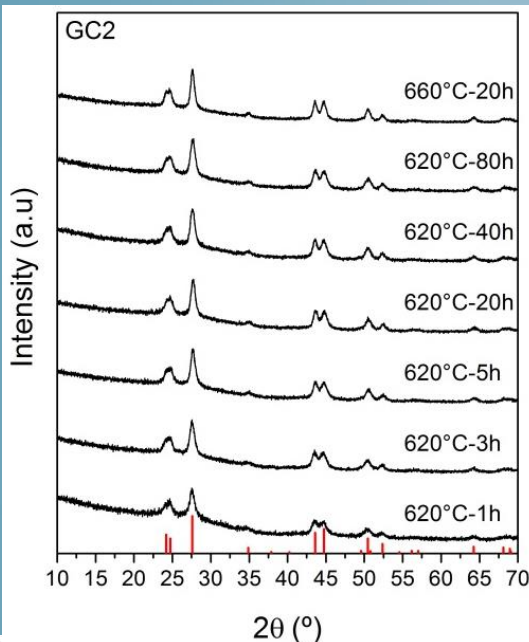


Increase of the dopant content results in a growth of the nanocrystal size

Sample	Size (nm)	Size (nm)
undoped	12.9	17.5
0.5Er ³⁺	13.2	20.1
0.5Er ³⁺ -2Yb ³⁺	16.8	23.1

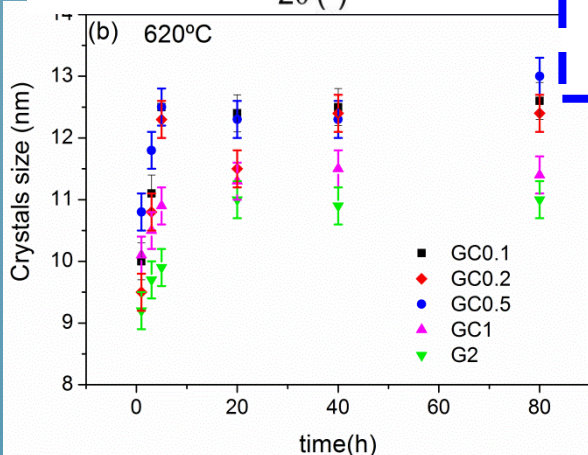
Structural characterization: XRD EAGLES

55Si-10La: Bulk and fibers

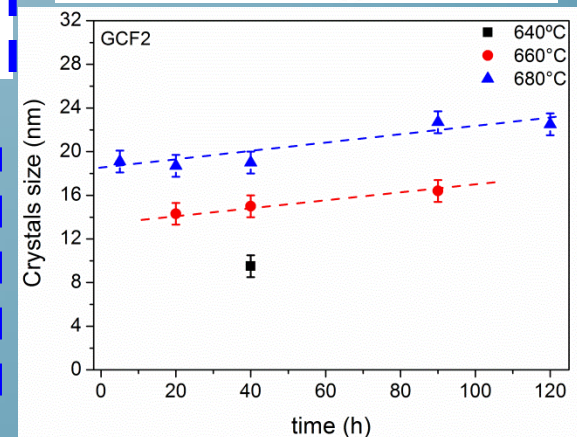
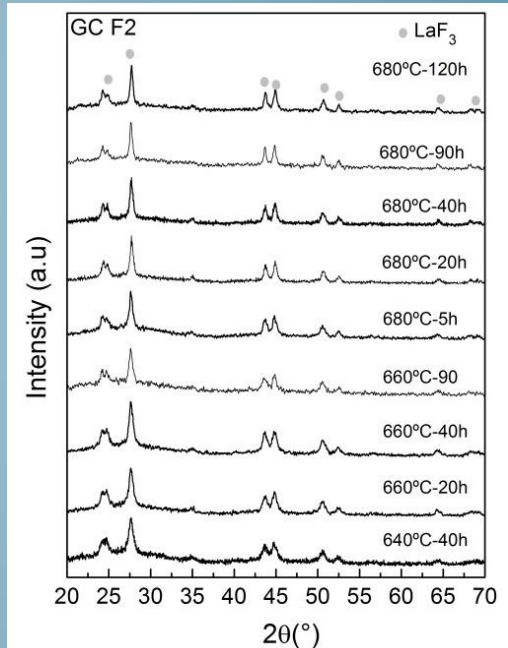


Only LaF_3 crystals precipitate in the glass matrix. The same behaviour is observed for all concentrations

Fibres treated at 620°C are still amorphous unless treated for $t > 120\text{h}$. A higher cooling rate in fibres causes crystallization delay



As Nd^{3+} content increases T_g increases as well and crystals size is smaller

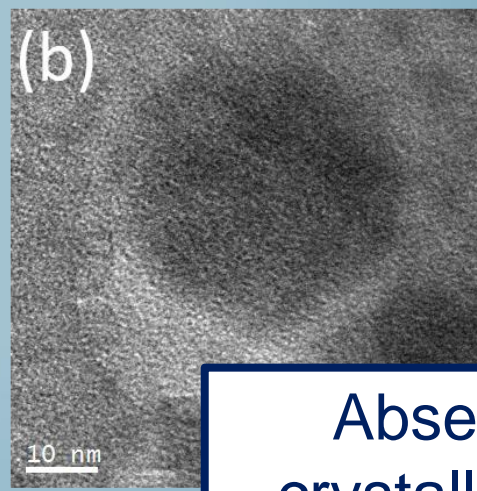
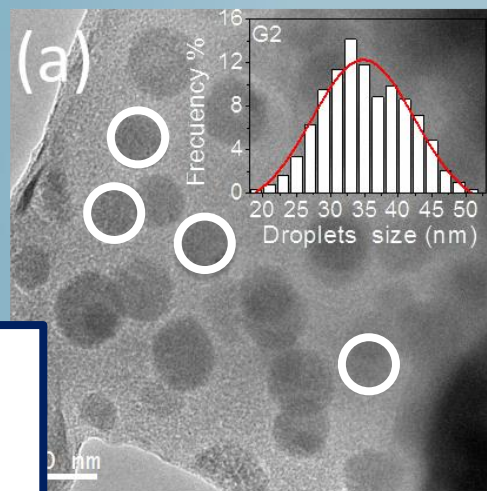


Structural characterization: HRTEM

Bulk Glass

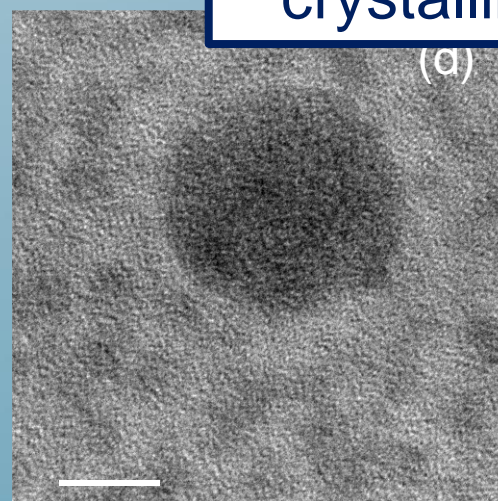
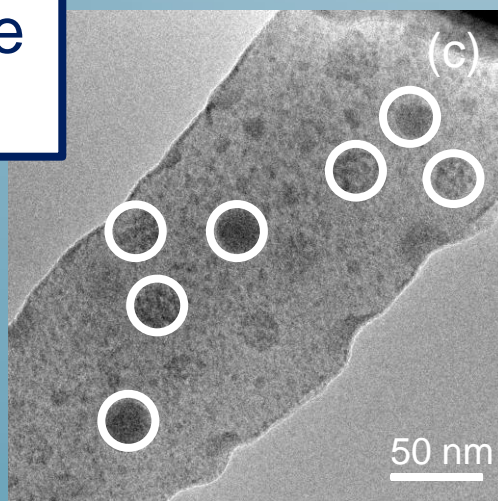
55Si 10La

Phase
separation
droplets in the
glass



Absence of any
crystalline structure

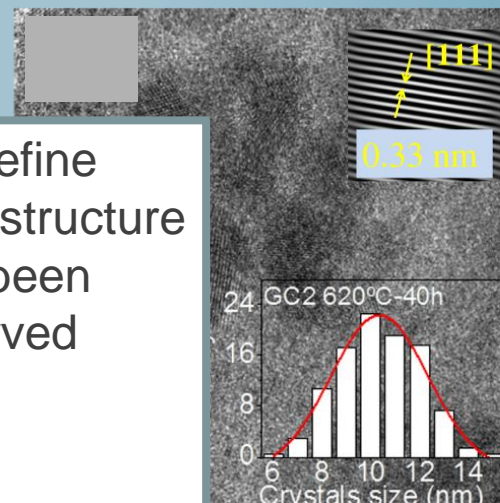
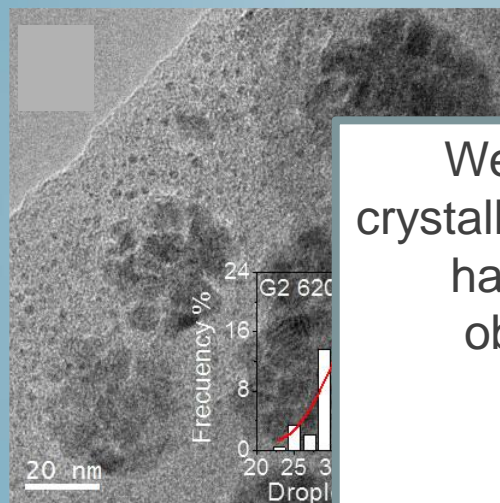
70Si 7Gd



Structural characterization: HRTEM

Bulk GC

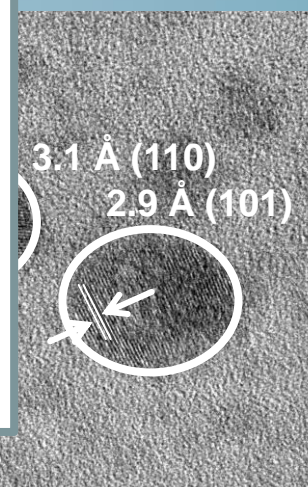
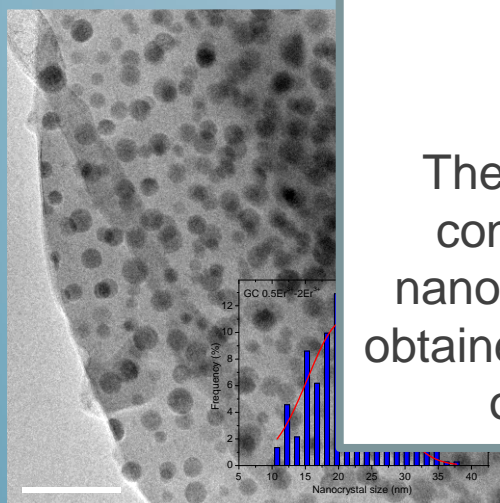
55Si 10La 2Nd³⁺
GC 620°-40h



Well define
crystalline structure
have been
observed



These results
confirms the
nanocrystal size
obtained by means
of XRD

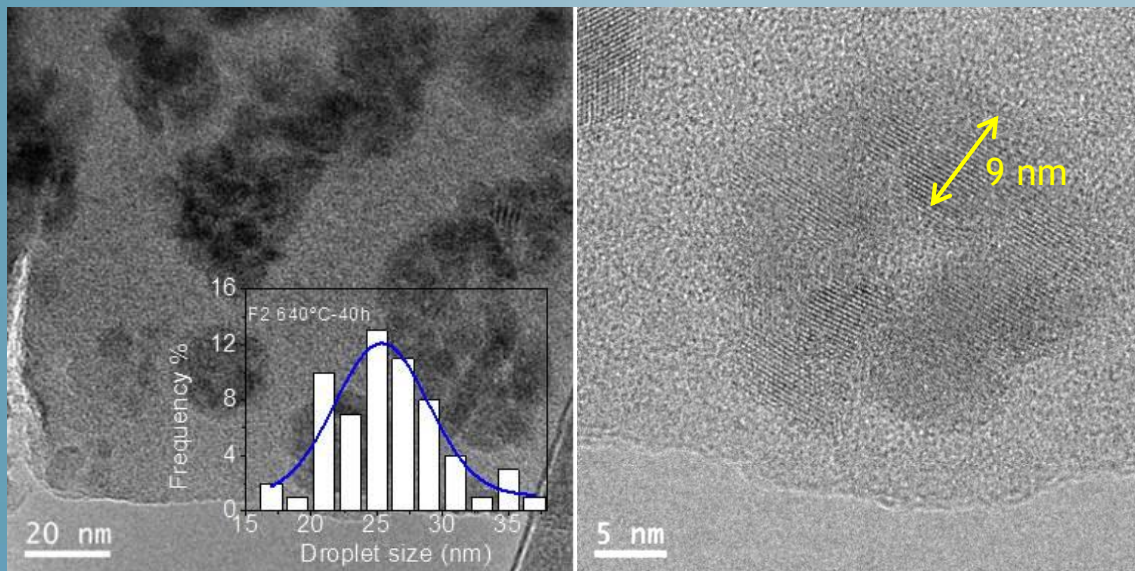


70Si 7Gd GC
0.5Er³⁺-2Yb³⁺
580°C-120h

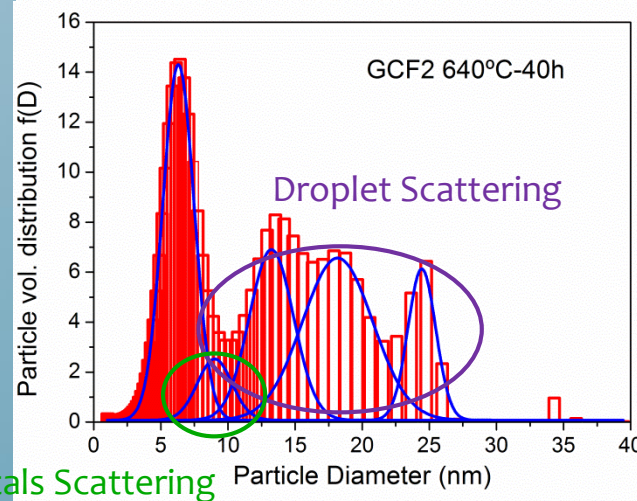
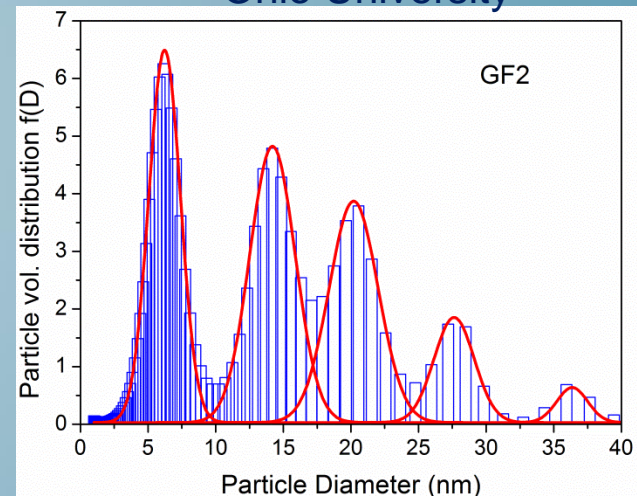
Structural characterization: HRTEM & SAXS

SAXS: Coll. Prof. Gang
Ohio University

Fibers GC



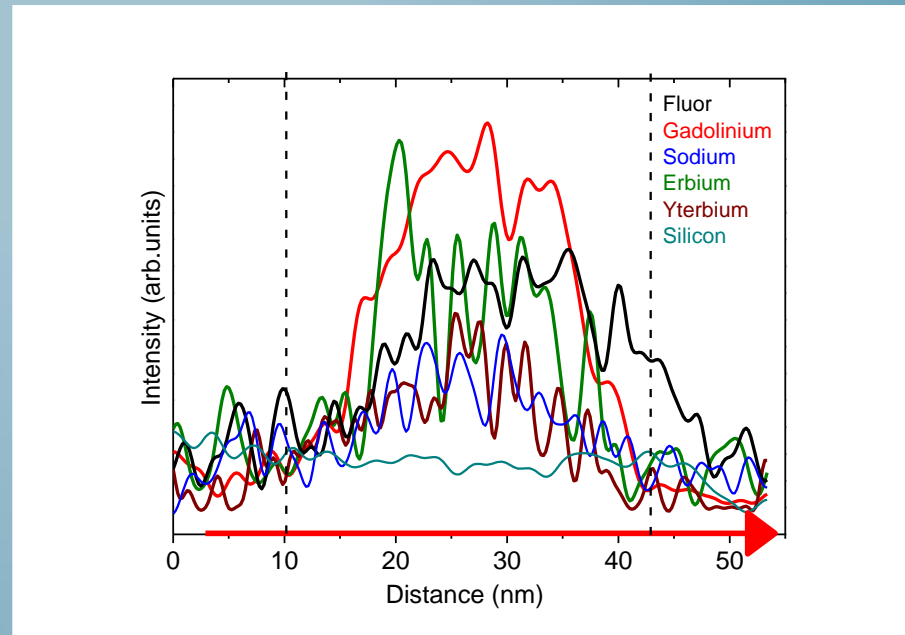
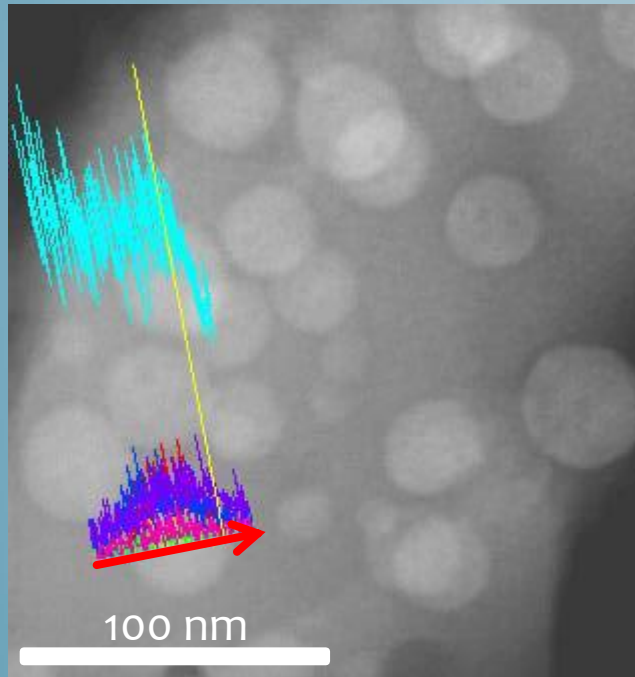
Smaller droplets are formed ~ 10% smaller compared to the bulk material but a narrower distribution is obtained



Structural characterization: STEM-EDX

Where are RE³⁺ ions?

70Si 7Gd GC



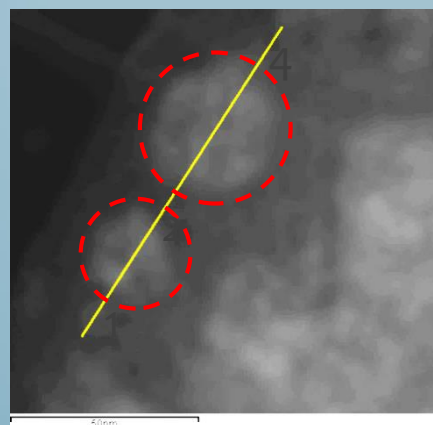
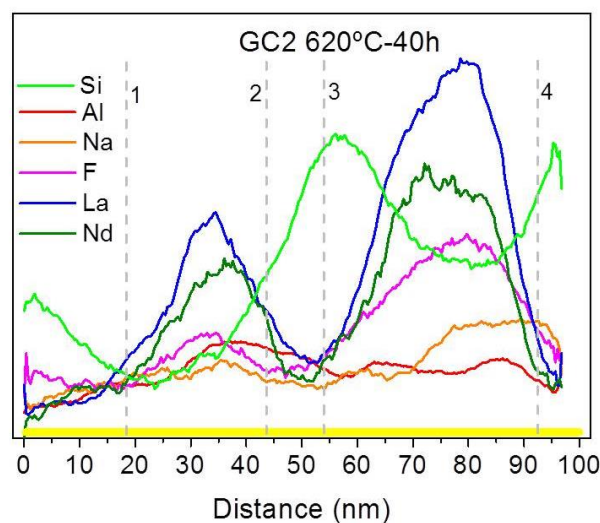
Gd, Na, F, Er and Yb are concentrate in
nanocrystals

Structural characterization: STEM-EDX

Where are RE³⁺ ions?

55Si 10La

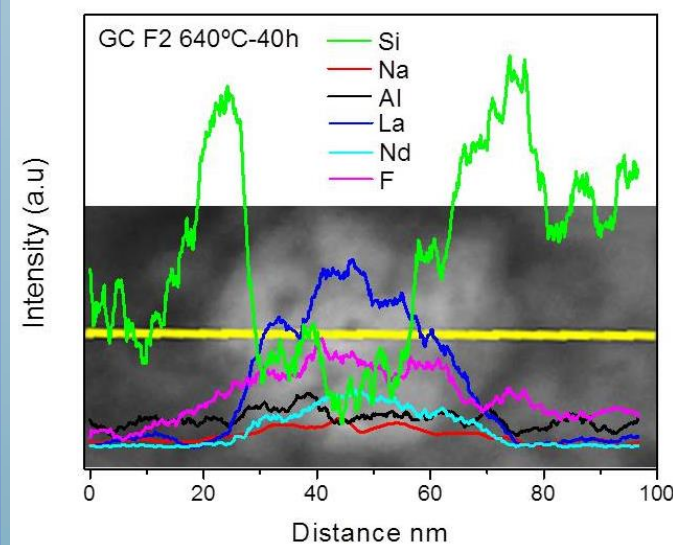
Bulk



EDX in STEM mode along 100 nm (two nearby droplets were analysed)

A viscous barrier enriched in Si and Al surrounds the droplet.
The droplet is clearly enriched in La, Nd and F.

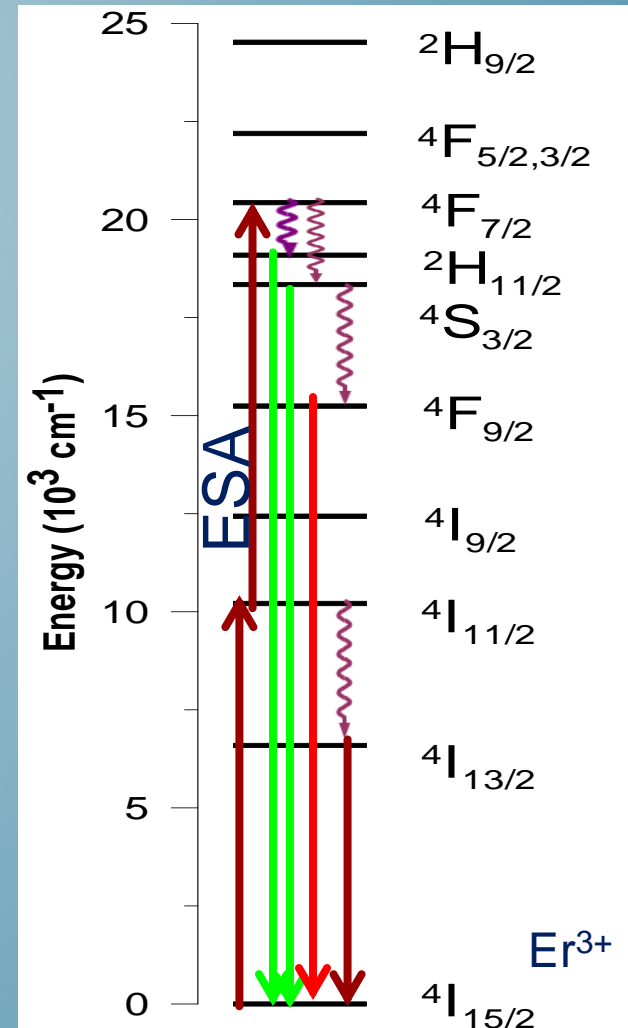
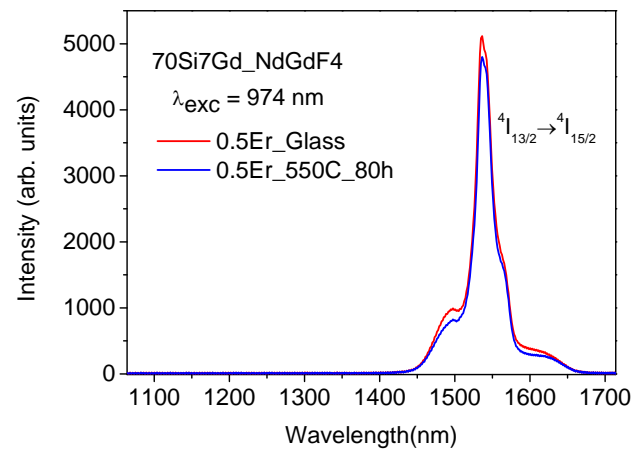
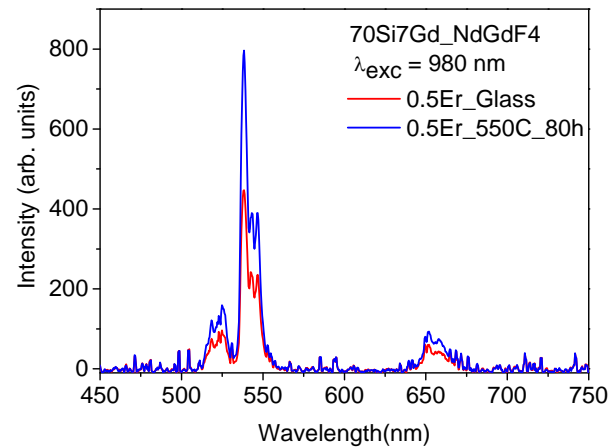
Fibres



From TEM and EDX the same structure and crystallization behavior as for bulk samples is observed.

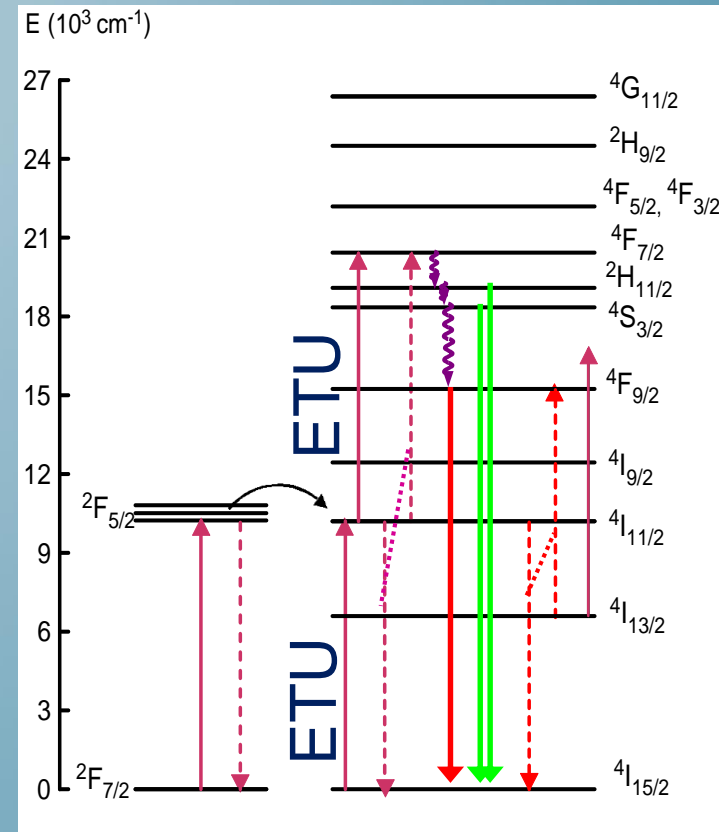
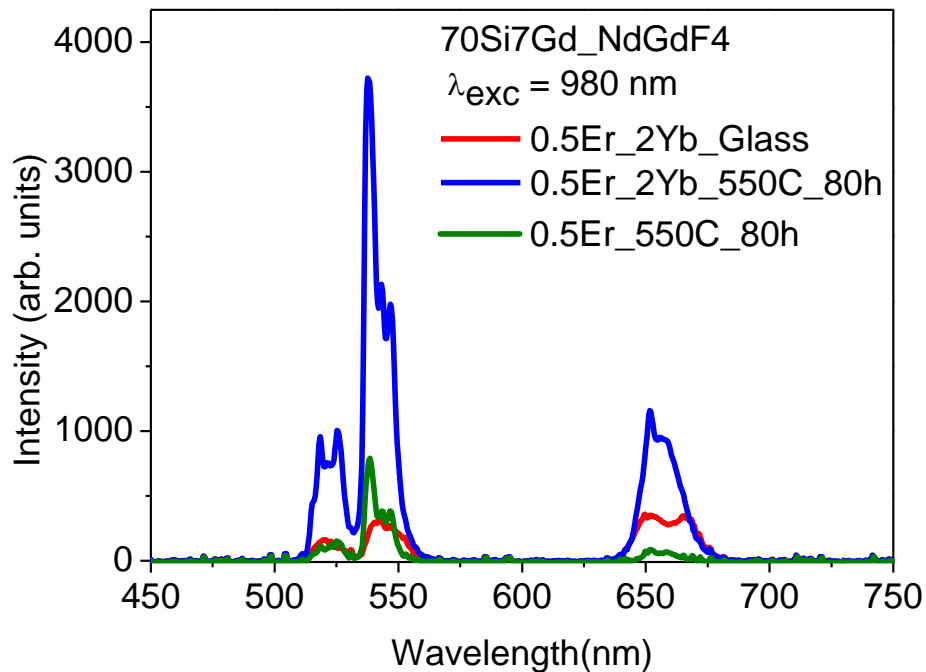
70Si 7Gd: 0.5Er³⁺

Up-conversion
(UC)



Emission spectra

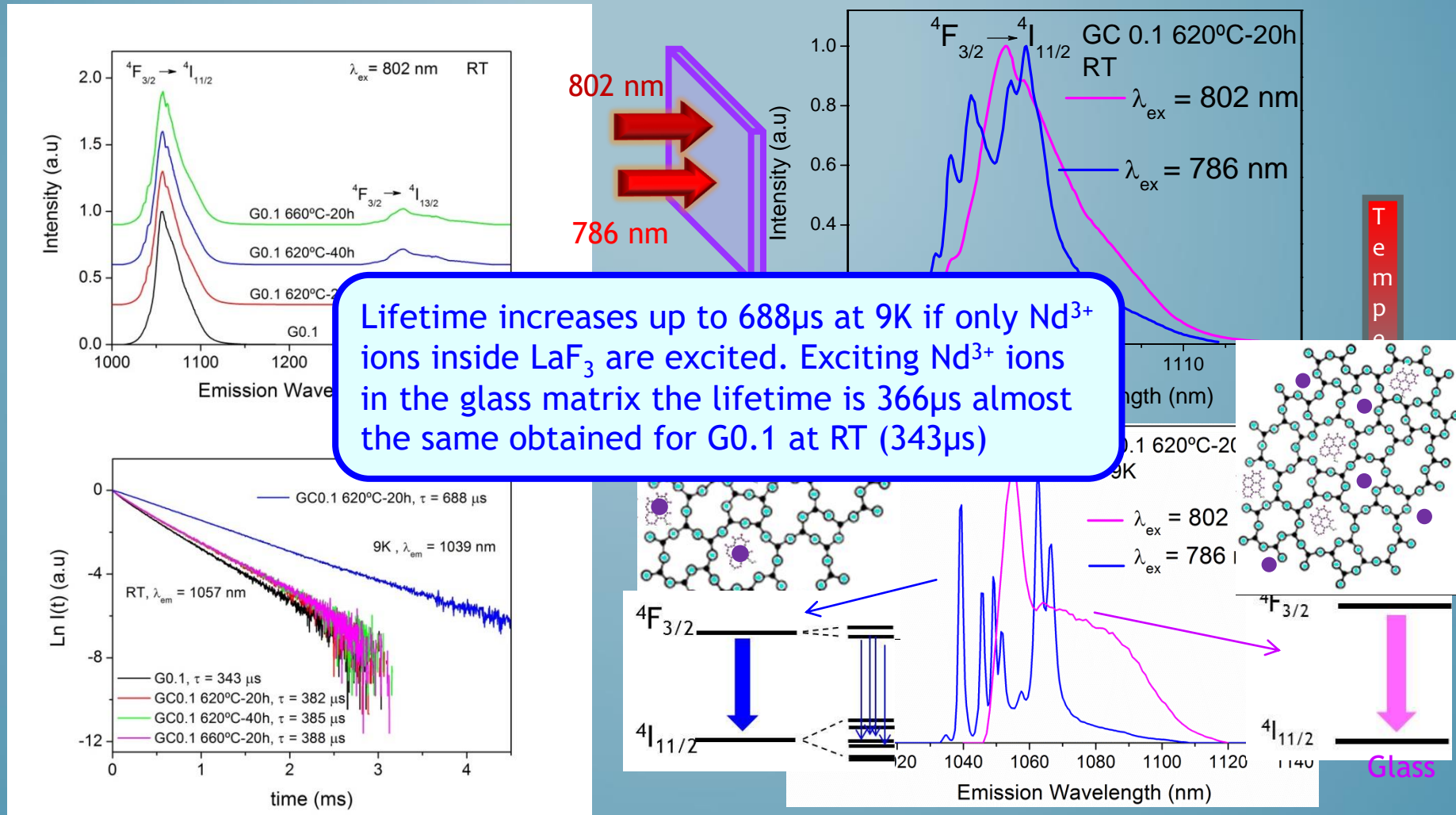
70Si 7Gd: 0.5Er³⁺-2Yb³⁺



Optical Characterization

PL and lifetime

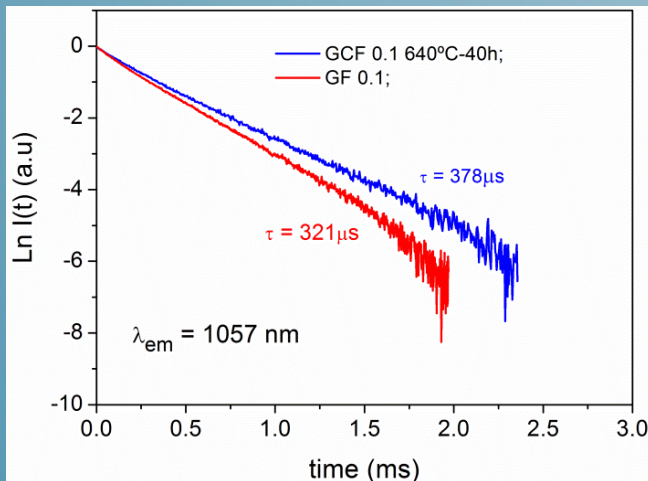
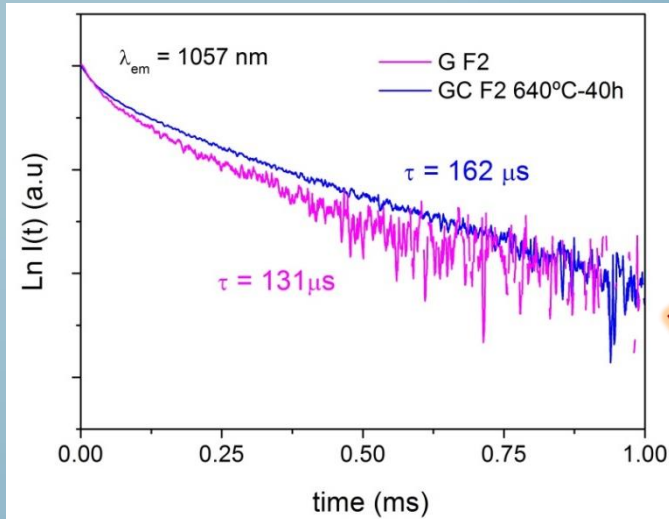
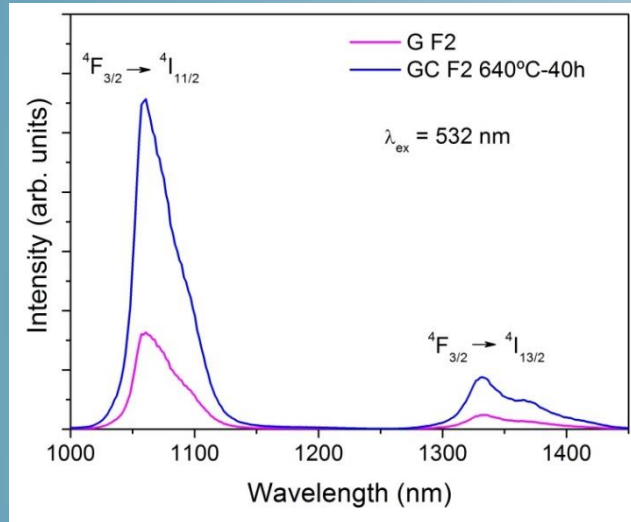
55Si 10La Bulk



Optical Characterization PL and lifetime

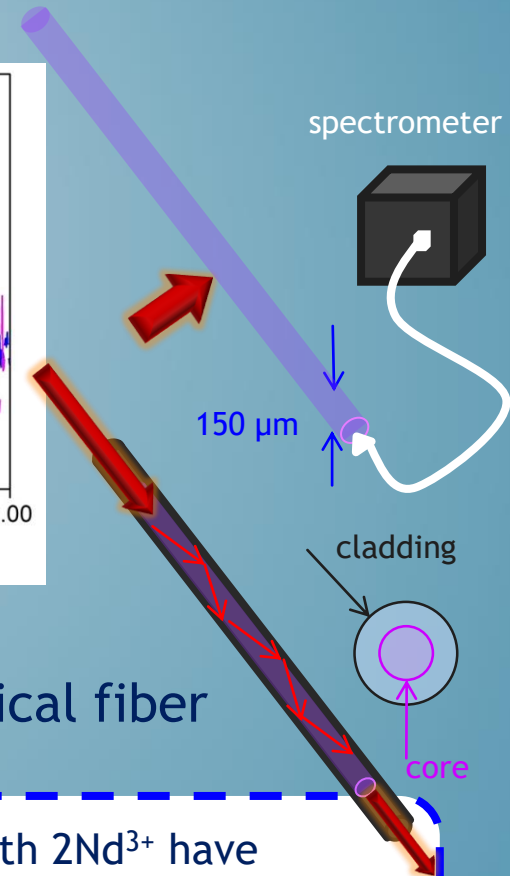
EAGLES

55Si 10La fibers



Next step: GC optical fiber

Glass and GC Fibers doped with 2Nd^{3+} have lifetimes of $131 \mu\text{s}$ and $162 \mu\text{s}$, respectively, very similar to the bulk Glass ($140 \mu\text{s}$) and GC ($180 \mu\text{s}$). The same behaviour occurs for 0.1Nd^{3+}



Conclusions

- *β -NaGdF₄ or LaF₃ oxyfluoride glass-ceramics have been successfully obtained by melting quenching method.*
- *After an adequate heat treatments, Hexagonal β -NaGdF₄ or LaF₃ nanocrystals with a mean size ranging from 8-30 nm have been obtained.*
- *According to the HRTEM results, crystallization mechanism occurs through regions of (Na,Gd)- or La- and Si phase separation in the glass, from which the fluoride crystals grow during heat treatment.*
- *By means of STEM-EDX we confirmed that the RE ions are effectively incorporated into the monocrystalline phases.*
- *Preliminary optical studies confirm that RE ions are partitioned into the nanocrystalline phase. Moreover, energy transfer processes between Er³⁺-Yb³⁺ ions or selective excitation of Nd³⁺ ions have been observed in NaGdF₄ and LaF₃ GC, respectively.*
- *Structural and optical properties of bulk and fibers in LaF₃ are very similar and preliminary studies of fibers can be made on bulk materials. For that, the best content was 0.1 Nd³⁺ ions.*



EAGLES



Prof. Gang Chen and
coll.



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MAT2013-48246-C2-2-P
FPDI-2013-16895



Thank you for you attention !

