



Nanoparticles containing glasses prepared using direct doping method

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As part of CoACH-ETN program



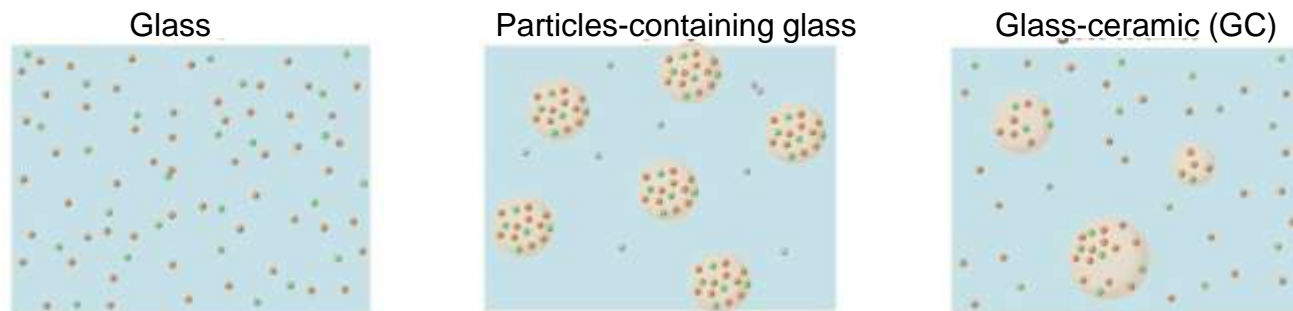
Outline

Main objective: To improve the spectroscopic properties (absorption, emission and lifetime) of Er^{3+} -doped phosphate glasses

- To control the site of the rare-earth

- 1) Particles-containing glass obtained by doping directly the glass with the nanoparticles
- 2) GC obtained by heat treating the glass leading to in-situ particles growth in glass

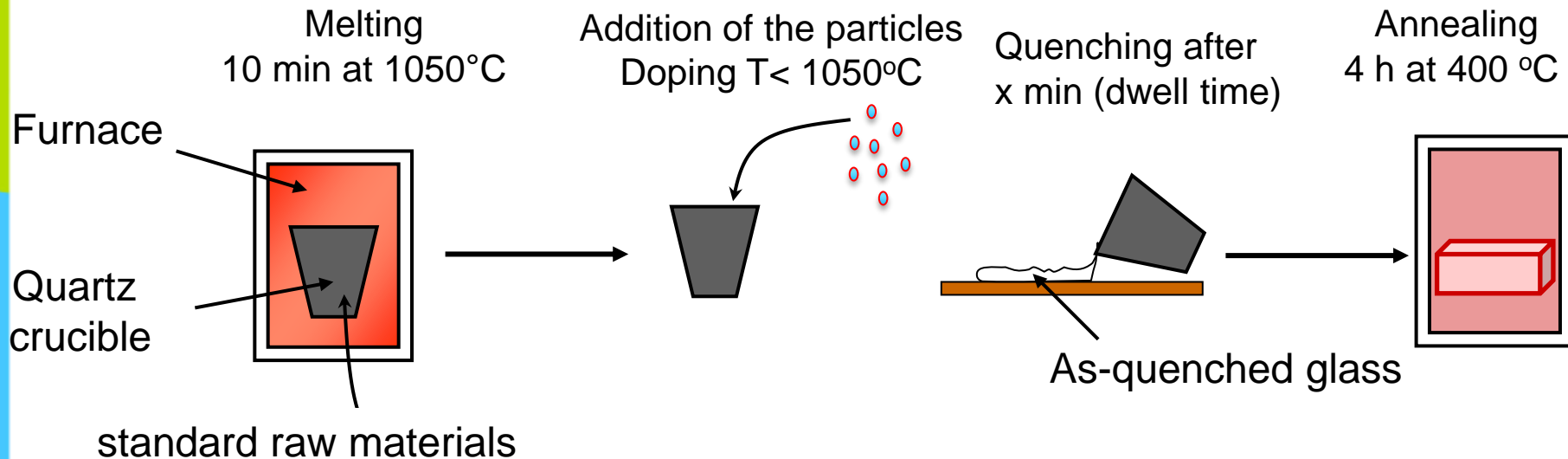
Schematic diagram of RE distribution in various glasses*



Direct particles doping method

Glass system: $50\text{P}_2\text{O}_5$ - $10\text{Na}_2\text{O}$ - 40SrO (mol%)

Modified melting process



To balance the survival and dispersion of the particles, 2 parameters to optimize:

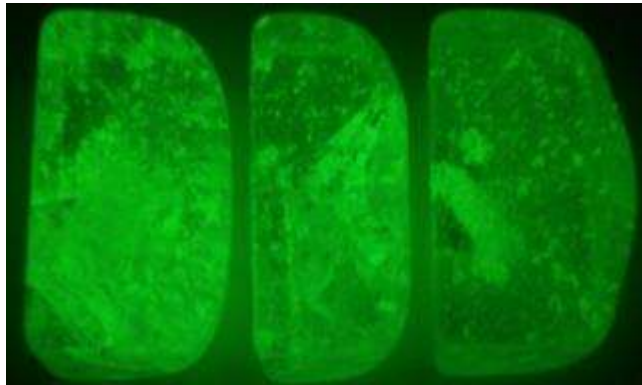
- 1) doping temperature
- 2) dwell time before quenching the glass



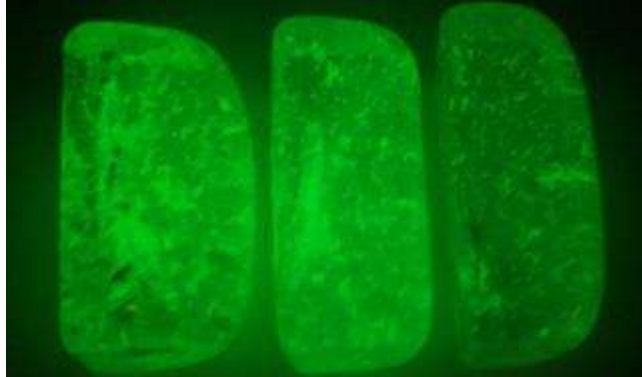
Optimization of the direct doping process

Temperature
975°C 1000°C 1025°C

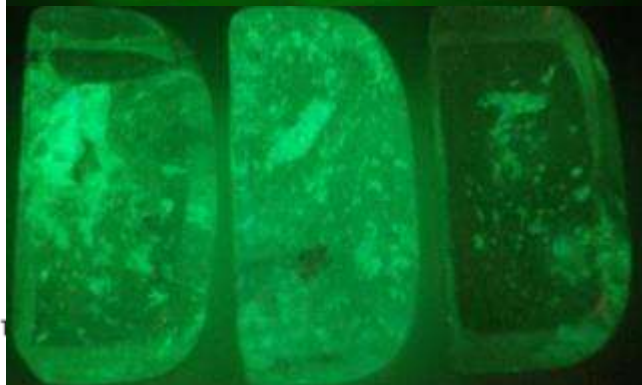
3min



5min



10min



Analysis of the images using ImageJ,
Java-based image processing program

Melting condition		
Dwell time	Temperature (°C)	SdtDev
3min	975	10
	1000	11
	1025	9
5min	975	10
	1000	8
	1025	6
10min	975	20
	1000	12
	1025	11

*Glass with more uniform PL = SdtDev as
small as possible*

Based on ImageJ analysis and on the
luminescence properties of the glasses

Optimized doping temperature = 1000°C
Optimized dwell time= 5min

Particles containing glasses

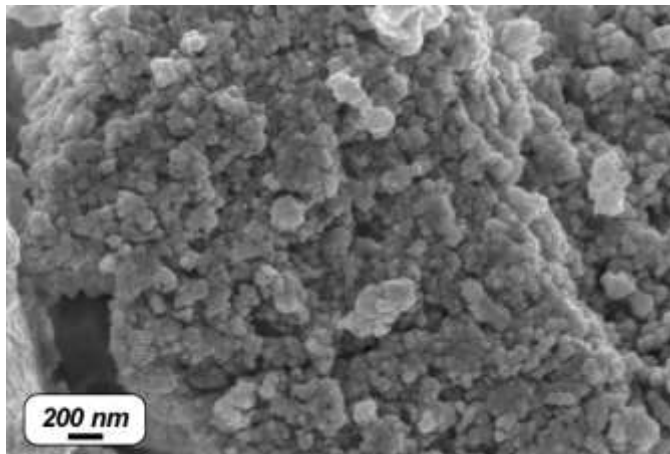
Glass matrix: $50\text{P}_2\text{O}_5$ - $10\text{Na}_2\text{O}$ - 40SrO (mol%)

Particles: Er: Al_2O_3 , Er: TiO_2 and Er,Yb:YAG ($\text{Y}_3\text{Al}_5\text{O}_{12}$)

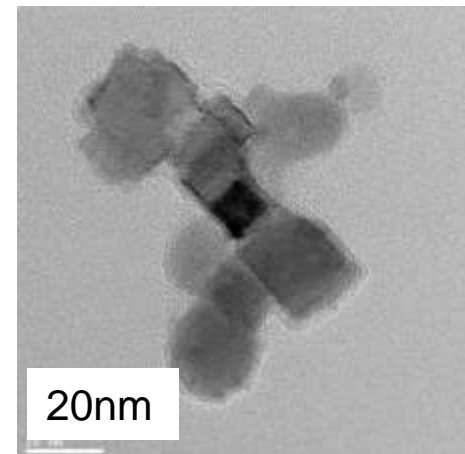
YAG NPs also prepared with a layer of silica with a thickness of 5 (YAG5) and 25nm (YAG25)

All particles are thermally stable at 1000°C , the doping temperature

SEM image of Er: TiO_2

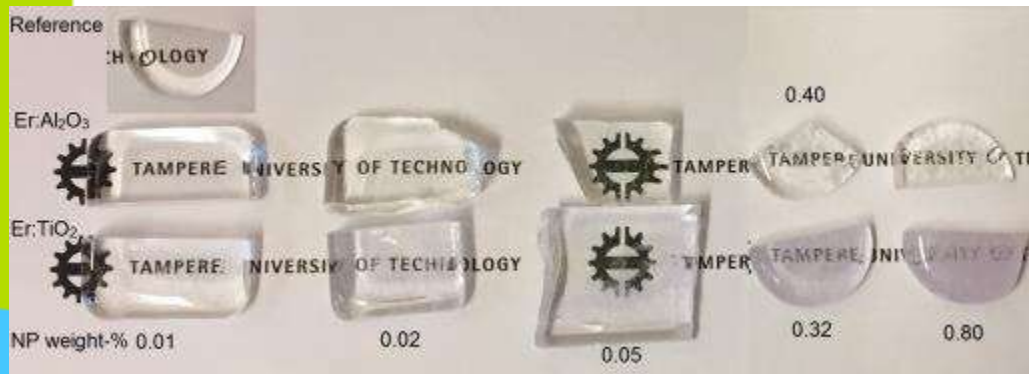


TEM image of YAG5



Er: Al₂O₃ & Er:TiO₂ containing glasses

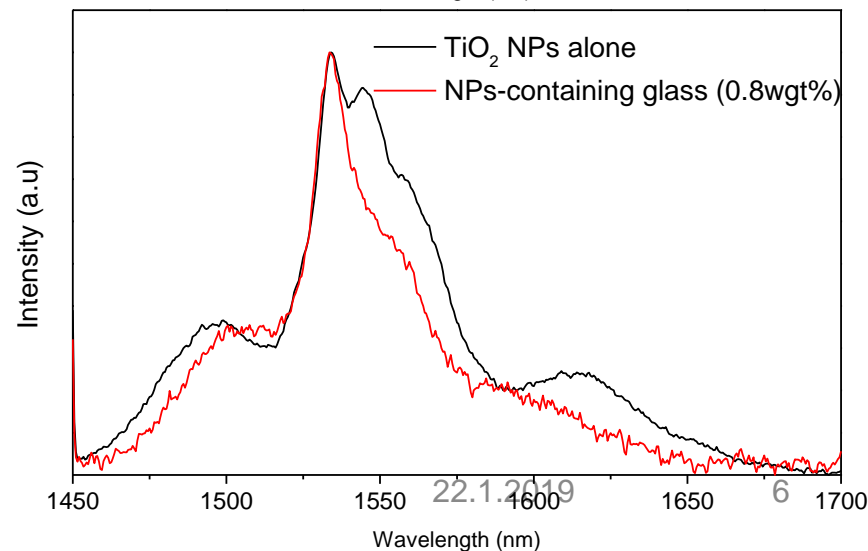
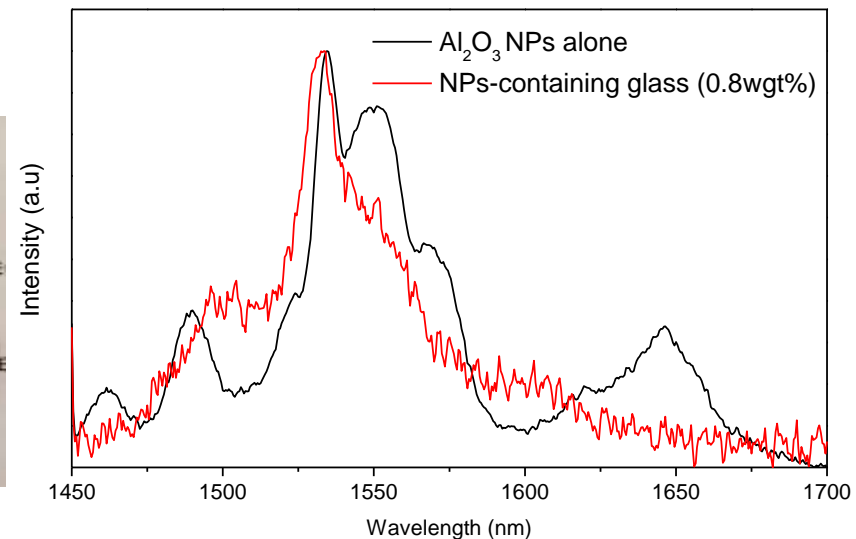
Picture of prepared glasses along with the weight-% of added NPs



Emission band of the NPs is different after being added in the glass
→ Er³⁺ in amorphous network

in agreement with the increase in the Er³⁺:⁴I_{13/2} lifetime after adding the NPs in the glasses

Normalized emission spectra
($\lambda_{\text{exc}}=976\text{nm}$)



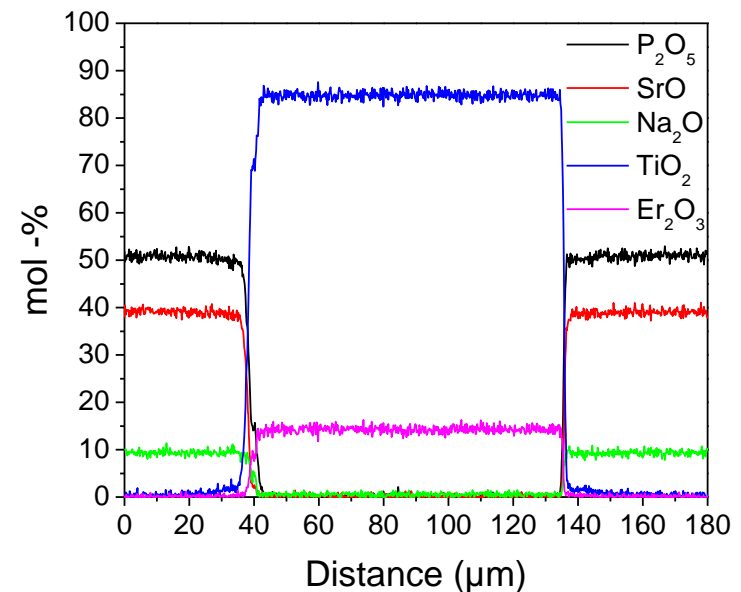
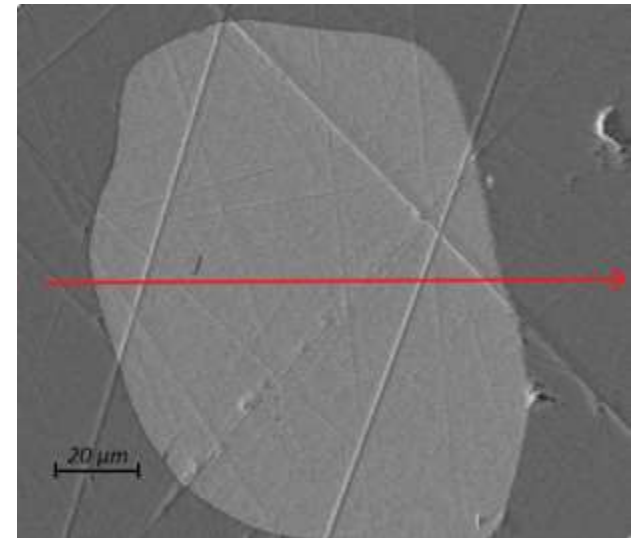
SEM/EDS

Er:TiO₂ –particles were found very easily with sizes up to ~100 μm.

Composition of the glass in agreement with the target composition of 50P₂O₅–40SrO–10Na₂O (mol%).

The particles were compositionally stable in the center

A small amount of NPs does survive during the glass preparation



YAG-containing glasses

Picture of YAG-containing glasses
(1.25wgt% of YAG)



with 5nm silica layer

with 25nm silica layer

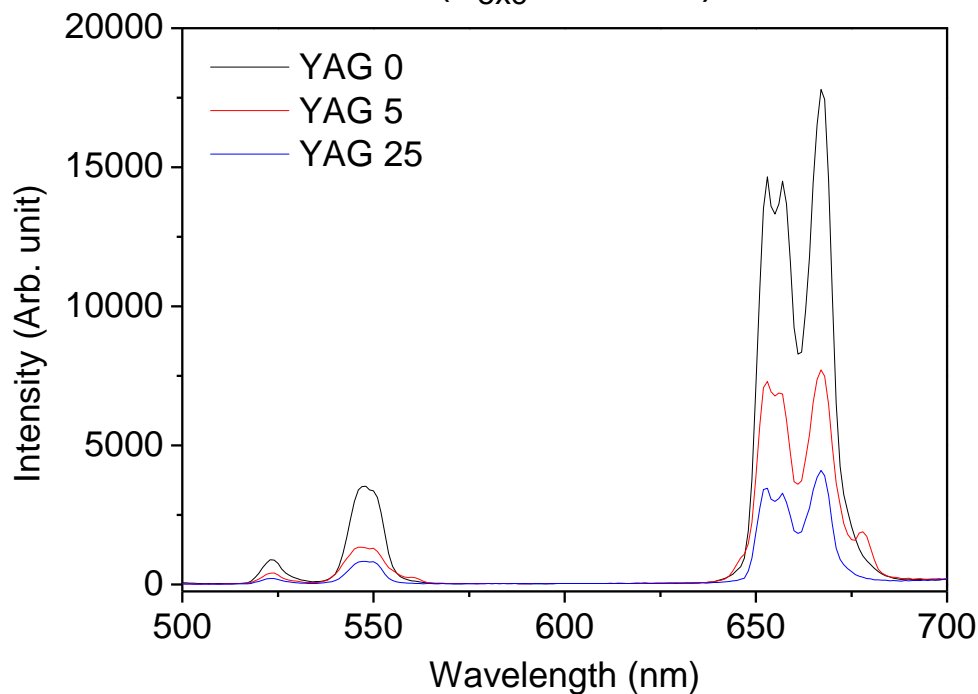
1.25w% of YAG
=
0.0038mol%Er₂O₃ –
0.035mol%Yb₂O₃

Agglomerates from the drying of the YAG-containing solutions.

Successful preparation of
phosphate glasses with
upconversion
while using small amount of
RE

Stronger upconversion from
YAG0-containing glasses

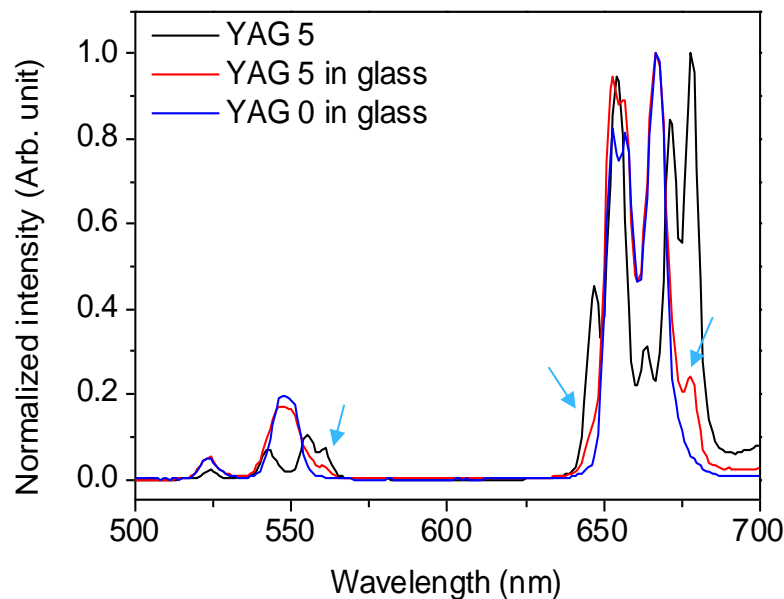
Upconversion spectra
($\lambda_{\text{exc}}=980\text{nm}$)



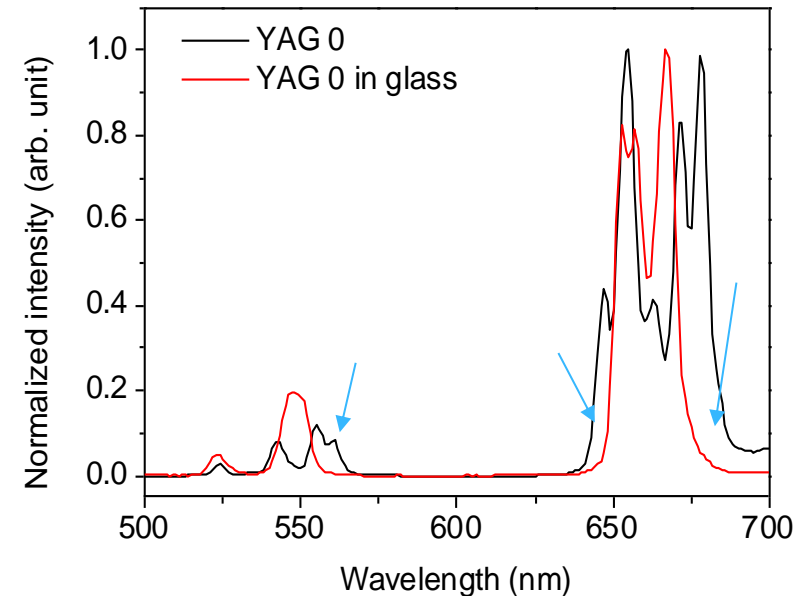
YAG-containing glasses

Different emission shape when NPs
are added in the glass
→ Er^{3+} environment modified.

Normalized upconversion spectra
($\lambda_{\text{exc}} = 980\text{nm}$)



Normalized upconversion spectra
($\lambda_{\text{exc}} = 980\text{nm}$)



Residual peaks of YAG in YAG5-containing glass.

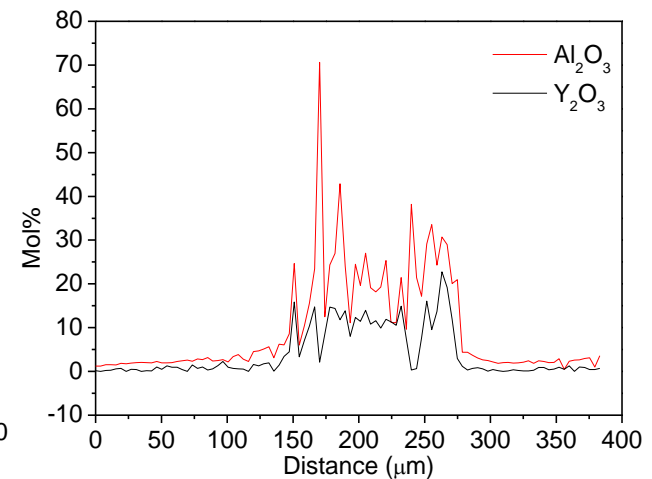
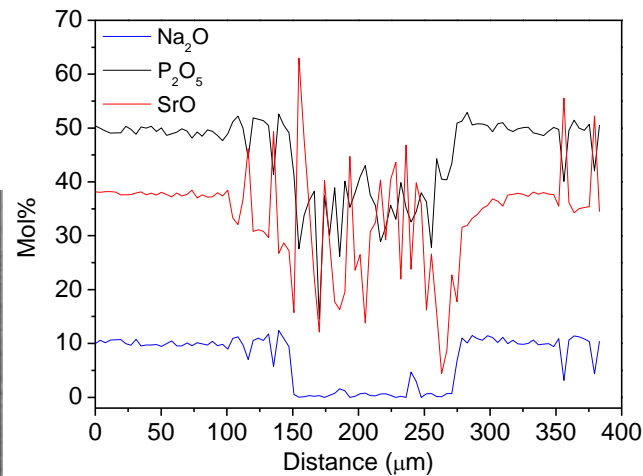
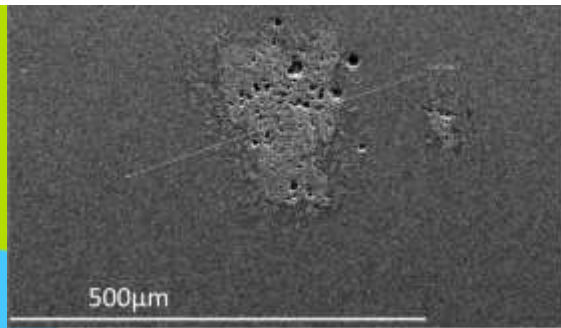
→ **Partial survival of YAG NP structure** in YAG 5

→ Thin layer of SiO_2 seems to have an impact on the survival of the particles.

YAG-containing glasses

Composition analysis along the line
(mol% in oxide)

SEM picture of an
agglomerate found in
YAG0-containing glass



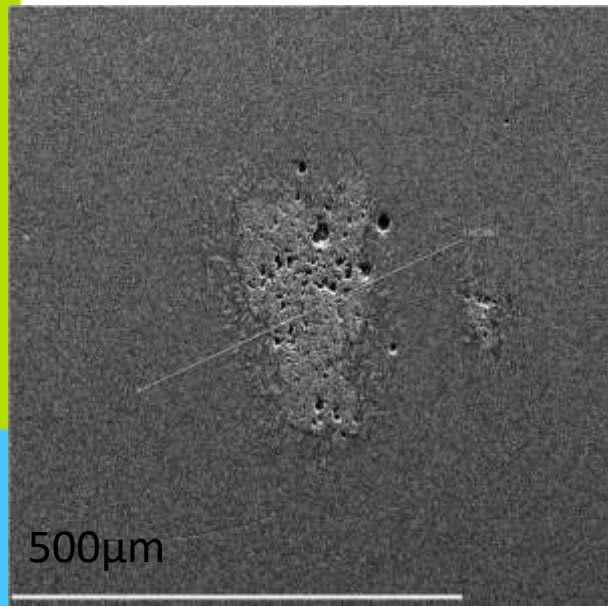
Composition of the glass in
accordance with the theoretical one
Sr-rich crystals seen around the
particles

Al₂O₃ and Y₂O₃ detected
(SiO₂ in the YAG5 and
YAG25 containing glasses)

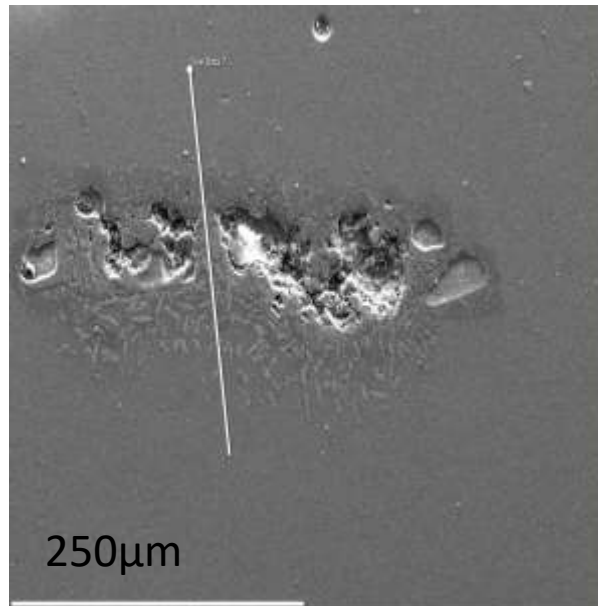
From the upconversion measurement and the composition analysis,
no doubt about the survival of the YAG NPs BUT with a different crystalline
phase.

YAG-containing glasses

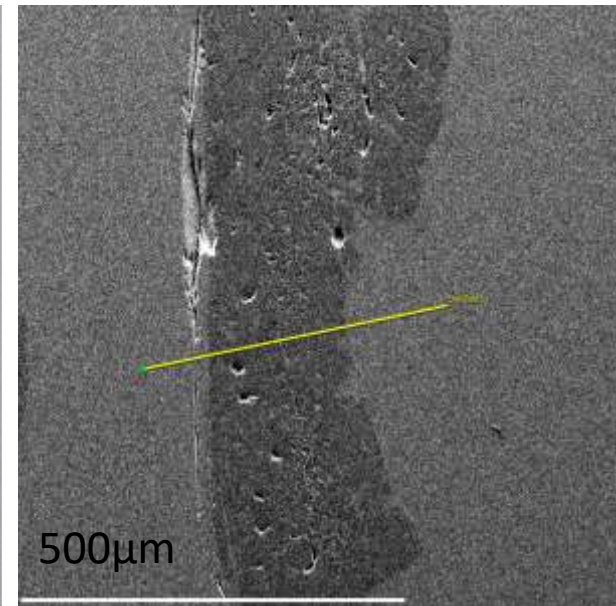
YAG0-containing glass



SEM picture of an agglomerate found in
YAG5--containing glass



YAG25-containing glass



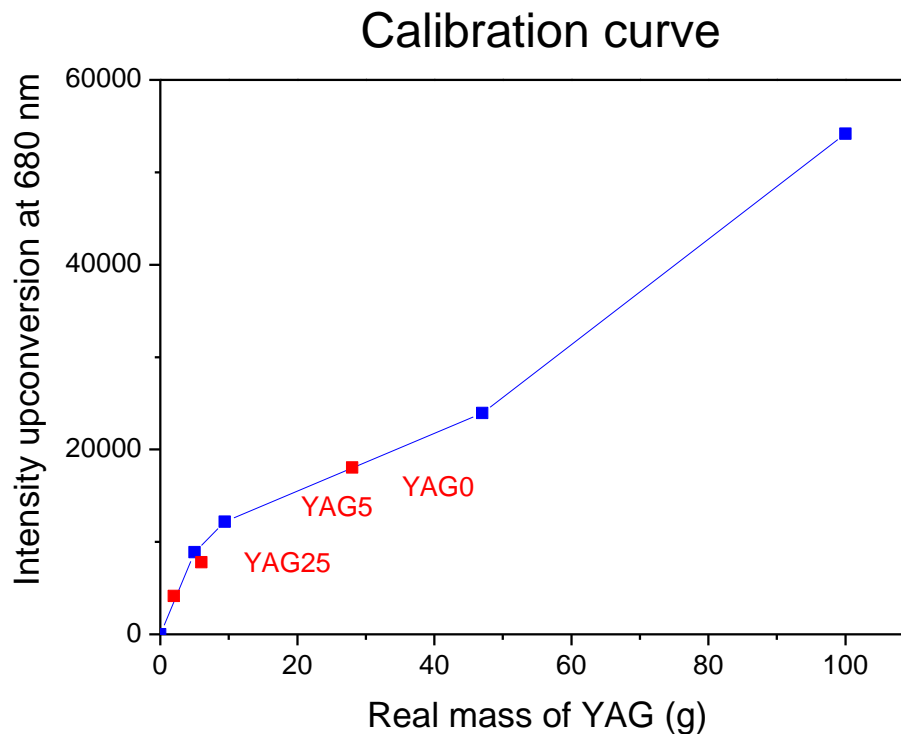
No crystals seen around the particles when the YAG are surrounded by a thick SiO_2 layer (25nm-YAG25)

Formation of crystals thought to be due to the diffusion of Al in the glass



YAG-containing glasses

Samples with known amount of YAGs were used to estimate the % of YAGs survival in the glasses



30-35 wgt% of YAGs survive during the glass processing

SiO₂ layer seems to increase the % survival of the YAGs
(To be confirmed)



Conclusion

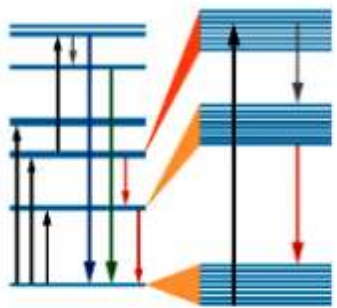
Possible to prepare particles containing glasses using direct doping method

- Upconversion can be obtained from a phosphate glass which contain 0.0038mol%Er₂O₃ – 0.035mol%Yb₂O₃ using YAGs. YAGs are supposed to survive but in different phases (to be identified)

- Thermal stability of the particles is not the only parameter to consider for the survival of the particles. The corrosion behavior of the glass melt is an important parameter to consider in order to engineer new NPs-containing glasses



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